

Methane emissions from rice microcosms: The balance of production, accumulation and oxidation

ULRIKE BOSSE* & PETER FRENZEL**

Max-Planck-Institut für terrestrische Mikrobiologie, D-35043 Marburg, Germany (* Present address: Department of Natural Resource Sciences, McGill University, Macdonald Campus, 21,111 Lakeshore Road, Ste. Anne de Bellevue, QC H9X 3V9, Canada; ** Corresponding author: Tel.: +49-6421-178821, Fax: +49-6421-178809, e-mail: frenzel@mail.uni-marburg.de)

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Abstract. In rice microcosms (*Oryza sativa*, var. Roma, type japonica), CH₄ emission, CH₄ production, CH₄ oxidation and CH₄ accumulation were measured over an entire vegetation period. Diffusive CH₄ emission was measured in closed chambers, CH₄ production was measured in soil samples, CH₄ oxidation was determined from the difference between oxic and anoxic emissions, and CH₄ accumulation was measured by analysis of porewater and gas bubbles. The sum of diffusive CH₄ emission, CH₄ oxidation, and CH₄ accumulation was only 60% of the cumulative CH₄ production. The two values diverged during the first 50 days (vegetative phase) and then again during the last 50 days (late reproductive phase and senescence) of the 150 day vegetation period. During the period of day 50–100 (early reproductive phase/flowering), the processes were balanced. Most likely, gas bubbles and diffusion limitation are responsible for the divergence in the early and late phases. The effect of rice on CH₄ production rates and CH₄ concentrations was studied by measuring these processes also in unplanted microcosms. Presence of rice plants lowered the CH₄ concentrations, but had no net effect on the CH₄ production rates.

Introduction

From preindustrial values of 0.7 ppmv, the atmospheric CH₄ mixing ratio has increased to currently about 1.7 ppmv (Raynaud et al. 1993; Kahlil & Rasmussen 1994). CH₄ is a very effective greenhouse gas (Lelieveld et al. 1993), and so its increase is cause for concern. Wetlands and irrigated rice fields are among the most important sources of atmospheric CH₄ (e.g. Cicerone & Oremland 1988; Wassmann et al. 1993). CH₄ emission from these sources is controlled by biological processes – microbial production and consumption – and by physical processes – diffusion, ebullition and ventilation. CH₄ production is a strictly anaerobic process. In contrast, CH₄ oxidation is almost exclusively an aerobic process. Only in marine sediments

and in some saline inland waters there is evidence for anaerobic CH₄ oxidation (Iversen & Joergensen 1985; Iversen et al. 1987).

The availability of O₂ is limited in wetland soils and in sediments, restricting aerobic CH₄ oxidation. The top soil of a few millimeters thickness is oxygenated, and a large percentage (about 90%) of the CH₄ diffusing through this layer is oxidized (Kuivila et al. 1988; Frenzel et al. 1990; King 1990; Conrad & Rothfuss 1991). Ebullition is often rapid and then CH₄ oxidation probably plays no role. Wetland plants constitute another effective pathway between the sediment and the atmosphere as their aerenchyma serves as a conduit for gases such as O₂ or CH₄ (Dacey & Klug 1979; Sebacher et al. 1985). While CH₄ will be transported into the atmosphere in this way, O₂ will be transported from the atmosphere downwards. O₂ may also leak from the plant roots into the surrounding sediment (Armstrong 1971) resulting in oxidized conditions in the rhizosphere (Crowder et al. 1987; Trolldenier 1988). Hence, favorable conditions for CH₄-oxidizing bacteria may be expected around roots of wetland plants, and indeed there is growing evidence that CH₄ oxidation in the rhizosphere is important in controlling the amount of CH₄ which is emitted from wetlands and rice paddies (King 1992; Bosse & Frenzel 1997). Measuring CH₄ oxidation in these systems is difficult, but it is a necessary step for understanding the mechanistic basis of CH₄ emissions, and CH₄ turnover in rice fields can be explained and modeled only if quantitative data are available. We measured these processes in rice microcosms over an entire vegetation period. To our knowledge this is the first attempt to compile such a process-based CH₄ balance in wetland rice.

Material and methods

Microcosms

Soil was taken from a rice field of the Istituto Sperimentale per la Cerealicoltura in Vercelli (Italy) in the spring of 1992 before flooding, and stored air dried. Both methanogenes and CH₄ oxidizing bacteria survive drying for prolonged periods (Le Mer et al. 1996; Mayer & Conrad 1990; Whittenbury et al. 1970). Field management and soil type at this site have been described (Holzapfel-Pschorn et al. 1986). The soil contained stubbles and rice roots from the last season. Before use the soil together with this organic matter was crushed and sieved to a diameter of <2 mm. Microcosms were constructed from Plexiglas[®]-tubes (inner diameter 6 cm), which were filled with water-logged soil to a depth of 14 cm and incubated at 25 °C under 2 cm of floodwater as described (Bosse & Frenzel 1997). They were planted with 4 rice seedlings each (*Oryza sativa*, var. Roma, type japonica) or left unplanted

as a control. Bulk soil had a density of 1.65 gfw ml^{-1} and a water content of $0.28 \text{ g H}_2\text{O (gfw soil)}^{-1}$.

Gas analyses

CH_4 and CH_3F were analyzed by gas chromatography with a flame ionization detector, O_2 was analyzed by gas chromatography with a thermal conductivity detector (Bosse & Frenzel 1997).

CH_4 concentration

To measure CH_4 concentration, one subcore was taken from the center of the microcosms, including one to two riceplants. Subcores were taken with a stainless steel corer (inner diameter 2.6 cm). The soil was pushed out with a piston and cut into 1–2 cm slices. The slices were immediately transferred into N_2 -flushed 150 ml flasks and diluted with the same volume of N_2 -bubbled demineralized water. The flasks were stoppered with rubber septa and flushed again with N_2 . CH_4 was extracted into the headspace by shaking for 2 min, and freshweight and dryweight of the soil were determined. By this method both porewater- and bubble- CH_4 was measured. This was shown using CH_4 probes (Rothfuss & Conrad 1994; Rothfuss et al. 1994) in the same microcosms, because with these probes it is possible to distinguish between porewater and gas bubbles. But some gas bubbles could also escape during the preparation. To measure total CH_4 , microcosms (soil volume ca. 0.4 l) were completely transferred into 2 l Erlenmeyer flasks containing 0.5 l degassed water. A Tedlar[®]-bag was connected to the flask to avoid pressure changes while the soil was transferred into the flask. The CH_4 was extracted into the headspace and freshweight and dryweight were determined.

CH_4 production

CH_4 production was measured in two ways: as anoxic fluxes (see next paragraph) and in soil slurries. For the slurries, slices and flasks with water were prepared as described. After transfer into the flasks, soil and water were mixed and root fragments picked out with forceps while the headspace was flushed with N_2 . The flasks were stoppered with rubber septa, flushed again with N_2 and incubated at 25°C in the dark. Before sampling the headspace for CH_4 , the flasks were shaken to equilibrate porewater- and headspace- CH_4 . CH_4 production in the soil was calculated from the increase of CH_4 in the headspace with time and from the dryweight of the soil slices, and expressed in $[\text{nmol (gdw)}^{-1} \text{ h}^{-1}]$. CH_4 production per area was calculated by summing up rates from $[\text{nmol (soil slice)}^{-1} \text{ h}^{-1}]$ to give $[\text{nmol subcore}^{-1} \text{ h}^{-1}]$, and by

converting this to $[\text{mmol m}^{-2} \text{ d}^{-1}]$ for the given production depth (14 cm) using the surface area of a subcore.

CH₄ emission and CH₄ oxidation

For measure emission, a chamber (volume 0.6–2.4 l) with a built-in fan was put over the microcosms into the water. Samples were taken with a gas-tight syringe through a butyl septum and the sampled volume was replaced with air. Fluxes were usually measured within 1 h ($n = 5\text{--}8$). Emission was calculated from the increase of CH₄ with time.

CH₄ oxidation was measured by two different methods. The first was to compare CH₄ emissions under oxic and anoxic conditions (also termed oxic and anoxic fluxes) and to use the difference between them as an estimate of CH₄ oxidation. For measurements of anoxic fluxes, the chamber was placed in the dark to prevent photosynthetic O₂ production, flushed with N₂ until O₂ was no longer detected and preincubated for ≥ 1.5 h before the flux was measured. The other method was the application of CH₃F. CH₃F inhibits methane monooxygenase, the key enzyme of CH₄ oxidation in methanotrophic bacteria. CH₄ oxidation is then determined from the difference between the fluxes with and without CH₃F. CH₃F was injected into the headspace to give a final mixing ratio of 1.6–2.4% and the chamber was preincubated for 3.5 h before the flux was measured. Anoxic fluxes and fluxes under CH₃F were usually measured within 2.5 hours ($n = 3\text{--}8$)

Dataset

CH₄ fluxes under an oxic atmosphere were measured from 22 planted microcosms. Eight of these were then treated with N₂ to measure anoxic fluxes and two with CH₃F to inhibit CH₄ oxidation. The 12 untreated microcosms and in addition 26 planted and 30 unplanted microcosms were used in the destructive experiments: Porewater CH₄ concentration was measured in 16 unplanted and in 15 planted microcosms, CH₄ production in 11 unplanted and in 10 planted microcosms, and CH₄ accumulation in 3 unplanted and 3 planted microcosms.

Results

The vegetation period of the rice plants in the microcosms was 150 days. The density of plants was high as compared with the field, but in contrast to field plants, tillers were only formed at the end of the vegetation period. Roots developed mostly between 30 and 60 days after planting. The flowering stage

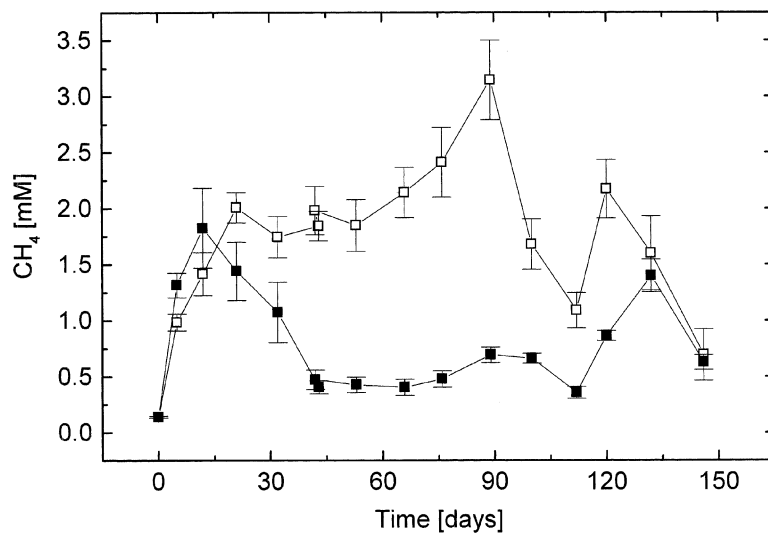


Figure 1. CH₄ concentration in planted (■) and unplanted (□) microcosms. CH₄ concentration was determined in each microcosm in different depths ($n = 7-11$) and then averaged. Concentrations could be higher than in CH₄-saturated water because both porewater and bubble-CH₄ are included. Each datapoint represents mean and standard error from one microcosm; standard errors are added as an empirical measure for the variation between the different soil layers.

was reached after 90 days. At this time, about 70% of the root biomass was located in the upper 3 cm of the soil, creating a root layer similar to that found in the field. The remaining 30% of the roots grew mostly along the walls and the bottom of the microcosms. A reddish tinge of the soil around the roots and on the roots themselves indicated Fe(III) plaques and thus sites where O₂ may have been released. Entrapped gas bubbles were observed frequently in the soil, especially in 1–2 cm depth in unplanted microcosms.

CH₄ concentration

Measurements on soil subcores showed that CH₄ concentrations were 2–4 times higher in unplanted than in planted microcosms during most of the vegetation period (Figure 1). The total CH₄ content, i.e. porewater- and bubble-CH₄, was also higher in unplanted microcosms (Table 1). A microcosm contained about 180 ml of porewater (calculated from freshweight, soil density and water content of the soil). At 25 °C, the maximum solubility of CH₄ is 1300 μm, which is equivalent to a total of 240 μmol CH₄ dissolved in the porewater of a microcosm. Comparing this value with the total CH₄ content (Table 1) shows that in unplanted and planted microcosms, at least 85% and 50% of the CH₄ was in the form of gas bubbles, respectively.

Table 1. CH₄ content in porewater and in gas bubbles of microcosms; one microcosm per measurements.

Age of microcosms [days]	CH ₄ content [$\mu\text{mol microcosm}^{-1}$]	
	Unplanted	Planted
0		658
20	n.d. ¹	1635
80	1631	546
130	1764	661

¹ not determined

CH₄ production

CH₄ production measured as anoxic fluxes are described in the next section. In slurry measurements of CH₄ production rates, the r^2 of the linear regression was ≥ 0.98 ($n = 4-7$) in 95% of all cases. Usually, rates were measured within 24 h. However, rates were linear for a wide variety of sampling frequencies. Sampling intervals between 1 and 24 h were tested by measuring CH₄ production as described, but with varying sampling frequencies. For larger sampling intervals, the procedure was modified. After measuring the initial CH₄ concentration at time 0 in a total of 40 flasks, at every time point after that, the CH₄ concentration was determined in 2 flasks which were discarded afterwards. Thus each flask was only agitated once after the initial measurement. CH₄ production was constant for 42 days. This showed that it was independent of the sampling frequency.

The average CH₄ production was very similar in planted and unplanted microcosms (Figure 2). CH₄ production rates from day 30–150 were $52.1 \text{ mmol m}^{-2} \text{ d}^{-1}$ (± 4.2 , 95%CL, $n = 16$, n_i = accumulated production of one profile). CH₄ production rates averaged over time for each depth were also very similar in planted and unplanted microcosms (data not shown). Finally, in both planted and unplanted microcosms CH₄ production rates at 0–3 cm depth were somewhat lower ($10.5 \pm 2.1 \text{ nmol gdw}^{-1} \text{ h}^{-1}$, 95%CL, $n = 50$) than in the soil layers below 3 cm ($14.0 \pm 1.1 \text{ nmol gdw}^{-1} \text{ h}^{-1}$, 95%CL, $n = 103$).

CH₄ emission and CH₄ oxidation

For all CH₄ emission rates, the r^2 of the linear regression was > 0.99 . Diffusive CH₄ emission rates under oxic conditions (oxic fluxes) are shown in Figure 3. The mean value was $26.3 \text{ mmol m}^{-2} \text{ d}^{-1}$ (± 5.7 95% CL, $n = 22$, range 1.8–54 $\text{mmol m}^{-2} \text{ d}^{-1}$). Diffusive CH₄ emission rates under anoxic conditions (anoxic fluxes) were higher than oxic fluxes, indicating CH₄ oxidation rates

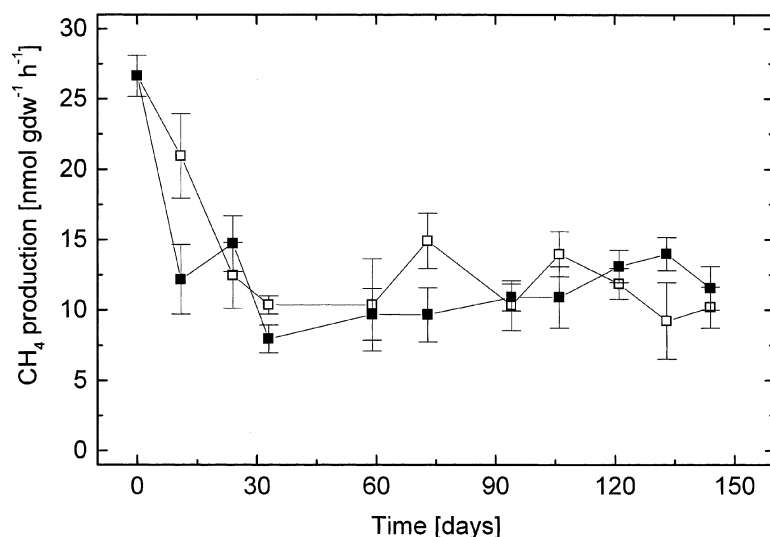


Figure 2. CH₄ production in planted (■) and unplanted (□) microcosms. CH₄ production was determined in each microcosm in different depths ($n = 4-10$) and then averaged. Each datapoint represents mean and standard error from one microcosm; standard errors are added as an empirical measure for the variation between the different soil layers.

between 1 and 19 mmol m⁻² d⁻¹ (Figure 4). The average percentage oxidation rate was 33% (± 9 SD, $n = 8$; range 17–43%) of the CH₄ production. If the inhibition experiments with CH₃F were included, the average oxidation rate was still 33% (± 9 SD, $n = 10$) (Figure 4). The percentage oxidation rates were constant with time ($r = 0.60$, $p > 0.05$, $n = 10$).

CH₄ balance

The seasonal CH₄ balance for a microcosm is

$$\text{production} = \text{accumulation} + \text{oxic emission} + \text{oxidation}.$$

For

$$\text{anoxic emission} = \text{oxic emission} + \text{oxidation},$$

this converts to

$$\text{production} = \text{accumulation} + \text{anoxic emission}.$$

We calculated the CH₄ production from the integrated production measured in slurry (Figure 2) and measured the accumulated CH₄ as total CH₄ content

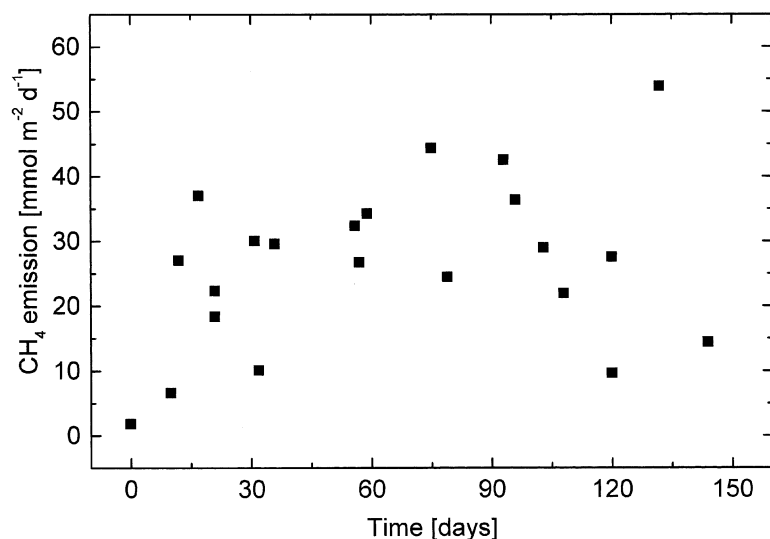


Figure 3. Diffusive CH₄ emission from planted microcosms under oxic conditions.

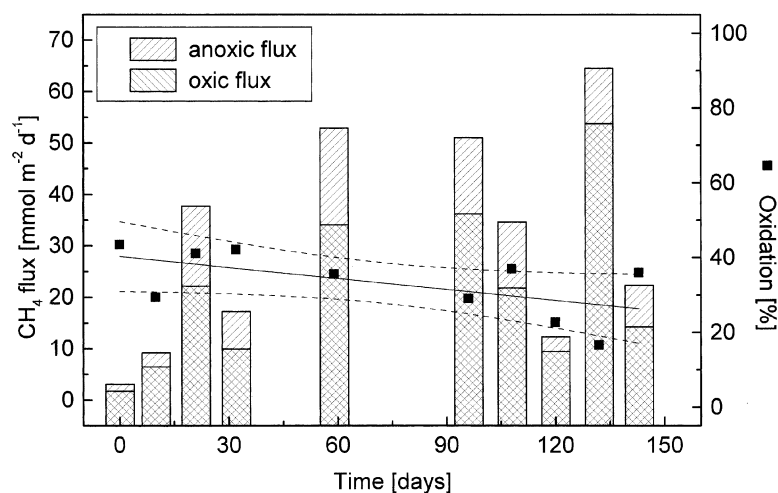


Figure 4. Diffusive CH₄ emission from planted microcosms under oxic and anoxic conditions. On each sampling date, the CH₄ emission from one microcosm was measured first under oxic (||||) and then under anoxic (////) conditions. All the fluxes under oxic conditions shown here are also shown in Figure 3. The difference between the two fluxes (where bars do not appear cross-hatched) was considered to be CH₄ oxidation. It was expressed as percentage of the anoxic flux (■, linear regression and 95% CL).

(Table 1). Diffusive anoxic emissions were measured directly (Figure 4). A seasonal balance was estimated from the measured values and is shown in Figure 5. First, values between measured data were interpolated linearly. This

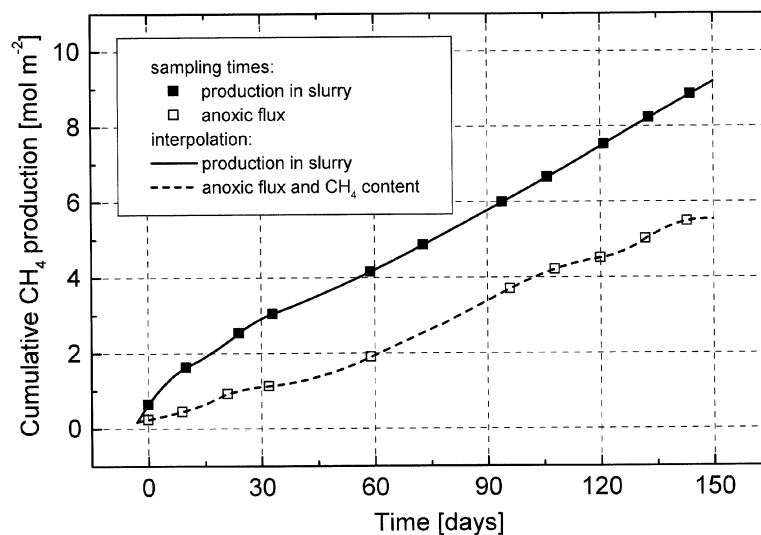


Figure 5. Cumulative CH₄ production in planted microcosms interpolated from slurry measurements (—■—) and from the diffusive anoxic flux plus the CH₄ content in the microcosms (—□—). For details of interpolation, see text.

was done with the CH₄ production rates, the CH₄ emission rates and the CH₄ content. The interpolated daily production rates were then summed up to give the cumulative production values shown as a solid line. The dashed line is composed of the interpolated daily anoxic emission rates summed up in the same way, and the CH₄ content. The daily CH₄ content was added directly, not as a cumulative value, as it was not a rate. Figure 5 shows that the sum of accumulation and diffusive anoxic emission represents only 60% of the production. The two lines diverge during the first 50 days and again after 100 days, but run more or less parallel in between.

Discussion

Processes

The results of the process-measurements mostly confirmed earlier studies. Planted microcosms contained less CH₄ than unplanted (Figure 1; Nouchi et al. 1994; Gilbert & Frenzel 1995). The CH₄ production rates were not affected by rice plants (Figure 2). In other studies, plant effects ranged from stimulation to inhibition of CH₄ production (Schütz et al. 1989; Sass et al. 1990; Butterbach-Bahl 1993; Frenzel et al. submitted). Many parameters, for example soil type, field treatment and rice variety probably influence CH₄

production. CH₄ emission rates (Figure 3) did not show a clear pattern. In field studies often more distinct peaks were observed, although the patterns vary considerably. It has been suggested that these peaks are the result of the sequential availability of straw, exudates and finally root biomass itself (Conrad 1993). Minoda et al. (1996), using $\delta^{13}\text{C}$ analyses, could show that the percentage of CH₄ derived from photosynthates had its maximum in July/August. Temperature can also influence CH₄ emissions (Butterbach-Bahl 1993). Our incubation conditions (constant light, constant temperature and no addition of straw) may thus have favored a simpler emission pattern. CH₄ oxidation was determined separately for the period of complete CH₄ balance (day 50–100) to be about 13 mmol m⁻² d⁻¹ or 25% of the CH₄ production. Our results indicate a constant percentage oxidation rate over the vegetation period. This was also found in other rice microcosms by Gilbert and Frenzel (1995), but is in contrast to field measurements in *Typha* (Lombardi et al. 1997). Percentage oxidation over different other studies varies a lot (Reeburgh et al. 1993). However, recent studies tend to give lower values (Denier van der Gon & Neue 1996) than older ones (Holzapfel-Pschorn et al. 1986). The absolute values in our experiment are within the range estimated in a rice field (5–15 mmol m⁻² d⁻¹, Holzapfel-Pschorn et al. 1985) and in other microcosms (mean of 5 mmol m⁻² d⁻¹, Gilbert & Frenzel 1995).

Balance

CH₄ turnover could be balanced for a 50-day period, but a complete balance was not possible (Figure 5). Two methods were used to determine the CH₄ production: measuring in slurry and measuring anoxic fluxes. For a considerable part of the vegetation period, CH₄ production was higher if measured in slurry. Schipper and Reddy (1996) reported similar problems with point measurements comparing slurry measurements with CH₃F fluxes for *Sagittaria lancifolia*.

For a balance, the reliability of the experimental procedures is very important. Production measurements in slurries may be biased in different ways, resulting in both over- or underestimation. Roots which were cut during the preparation could provide additional C-sources. Roots were therefore excluded from the slurries to prevent an overestimation of CH₄ production. However, exclusion of roots may instead result in underestimation of CH₄ production as roots give off organic compounds *in situ* which may stimulate CH₄ production. Rooted soil may contain oxidants such as O₂, NO₃⁻, SO₄²⁻, Fe(III) and Mn(IV) which can serve as alternative electron acceptors (Bacha & Hossner 1977; Trolldenier 1988; Frenzel et al. 1992; Wind & Conrad 1995; Arth et al. in press). Anoxic measurements will lead to a depletion of these electron acceptors, resulting in an overestimation of CH₄ production. How-

ever, from the depth profiles of CH_4 production, the CH_4 production of the main root zone (top 3 cm) was calculated. On average, this zone contributed only 15% to the overall CH_4 production. In contrast, the total difference between the production measured in slurry and the production measured as anoxic flux was larger (40%, including periods of correspondence). Finally, slurrying (i.e. homogenizing and suspending the soil) might have improved the substrate availability and thereby increased CH_4 production rates. However, we tested different sampling frequencies and found no effect on CH_4 production. We conclude that slurry measurements were unlikely to critically overestimate CH_4 production. Instead, CH_4 flux measurements under anoxic conditions may have underestimated CH_4 production.

If plants are incubated in a closed chamber, changes in temperature, humidity, light and CO_2 will occur and they will affect transpiration. However, in rice no correlation was seen between transpiration and CH_4 emissions (Nouchi et al. 1990; Butterbach-Bahl 1993), and in our own experiments, we measured linear increases of headspace- CH_4 without any indication of a chamber effect.

But flux measurements might be biased in other ways. During the first 50 days, anoxic diffusive fluxes probably underestimated CH_4 production, because the plants were still small and some CH_4 might have left the system as gas bubbles. Ebullition was not measured separately because the plant density was too high to install a sampling device. We saw one gas bubble flux in the very beginning (distinguished from the diffusive flux by its high CH_4 content), but we did not see any others. However, gas bubbles may be rare events and detection will be limited by the incubation time. The cumulative CH_4 production was 3700 mmol m^{-2} after the first 50 days, contrasting with a cumulative anoxic CH_4 emission (equivalent to oxic emission + oxidation) of 1100 and a total CH_4 content of 400 mmol m^{-2} . The difference of 2200 mmol m^{-2} is 24% of the seasonal cumulative CH_4 production of 9200 mmol m^{-2} . We assume that this CH_4 emitted as gas bubbles. In a field study, Butterbach-Bahl (1993) found that about 10% of the seasonal CH_4 emission emitted as gas bubbles during the first weeks.

CH_4 production from flux measurements would also be underestimated if the root system was not completely anoxic during the anoxic fluxes. Webb and Armstrong (1983) showed that when the green parts of 40 day-old rice plants were incubated under N_2 , roots stopped releasing O_2 after 15 min. A preincubation under N_2 of $\geq 1.5 \text{ h}$ should therefore be enough to ensure anoxic conditions even in the root system of the larger plants. Gas bubbles might have been a problem, because even a small amount of O_2 in a gas bubble represents a large reservoir compared to the amount dissolved in the surrounding porewater. This effect is difficult to assess. However, anoxic fluxes did agree with slurry measurements at a stage when the root system

was well developed and O₂ was probably present in the rooted layer (Frenzel et al. 1992). This indicates that O₂ in gas bubbles was not a problem and that CH₄ production was not underestimated in this way.

On the contrary, one argument against the N₂ technique is that an anoxic headspace favors CH₄ production in the soil. The anaerobic metabolism of the roots itself may stimulate root-associated CH₄ production, especially by releasing ethanol (Drew et al. 1994; Perata & Alpi 1994). Additional CH₄ production induced by the anoxic incubation would mean that the real divergence between the different estimates was even higher. An inhibitor like CH₃F would avoid this, and in our experiments, the average percentage oxidation rate measured under N₂ didn't change if the fluxes measured with CH₃F were included. But it was not used extensively because it may inhibit CH₄ production, too (Frenzel & Bosse 1996).

Finally, CH₄ production measured as CH₄ emission under anoxic conditions would be underestimated if anaerobic CH₄ oxidation took place. Anaerobic CH₄ oxidation has been reported mostly for marine and saline environments. In a Japanese rice soil it was suggested to occur in deeper soil layers coinciding with reduction of ferric iron (Kimura et al. 1992; Miura et al. 1992). However, anaerobic CH₄ oxidation would affect both slurry- and flux-measurements and so wouldn't help explain the difference between them.

Therefore, we think that the escape of CH₄ in gas bubbles offers the most plausible explanation for the observed discrepancy between CH₄ production measured in slurry and measured as CH₄ emission under anoxic conditions.

Between approximately day 50 and day 100, results from the slurry- and the flux measurements agreed well. The root system had reached its maximum biomass, providing a large aerenchyma for CH₄ to pass through. At the same time, escape of gas bubbles is probably mechanically impeded by the dense root mat. The potential for CH₄ oxidation in rhizospheric soil and roots was easily sufficient for the observed oxidation rate of 13 mmol m⁻² d⁻¹ (Bosse & Frenzel 1997).

After this period of good correspondence between flux measurements and slurry measurements, results diverged again after day 100. Gas transport through rice plants is mainly diffusive (Nouchi et al. 1990; Mariko et al. 1991; Denier van der Gon & van Breemen 1993). The capacity for transport may be limited, for example by the size of the root surface and especially at the transition zone between root and shoot (Armstrong 1979; Butterbach-Bahl 1993; Denier van der Gon & van Breemen 1993). A change in transport capacity due to morphological changes could affect gas transport under both oxic and anoxic conditions, but we found no information about this aspect of morphology during the late growth stage in the literature. However, we

observed more blackened, soft roots during this stage, especially after day 120, suggesting that plant-mediated transport might have become limited again.

Conclusion

Our results show that for a CH_4 balance in rice microcosms, CH_4 production, CH_4 emission and CH_4 oxidation were most important, while CH_4 accumulation was negligible. CH_4 production rates measured (a) in slurry and (b) as anoxic flux corresponded well for about one third of the vegetation period (day 50–100). Measuring CH_4 production with either of these methods, oxidation rates could then be calculated from the difference between the production and the oxic emission. The results confirmed that CH_4 oxidation is an important factor controlling CH_4 emissions, as 25% of the CH_4 production was oxidized.

However, a large amount of CH_4 was “lost” from the system during the vegetative phase (day 0–50) and some was lost during the late reproductive phase and senescence (day 100–150), most likely as episodic gas bubbles (ebullition). The total “loss” was 40% of the CH_4 produced. Under these conditions, CH_4 oxidation would be overestimated if it is calculated from the difference between oxic flux and production measured in slurry. We conclude that in principle it is possible to approach a balance, but that during the early vegetative phase and during senescence, CH_4 pathways seem to be quite variable and that more studies are needed to elucidate the role of gas bubbles and diffusion limitations in this system.

This study was done with microcosms, and because of the large heterogeneity in a field, a similar study would be difficult to do under field conditions. However, CH_4 emission rates were within the range observed in the field for this rice variety and this soil (Holzapfel-Pschorn & Seiler 1986; Butterbach-Bahl 1993). Thus it seems possible that the principle conclusions discussed here may also apply to the field situation.

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